

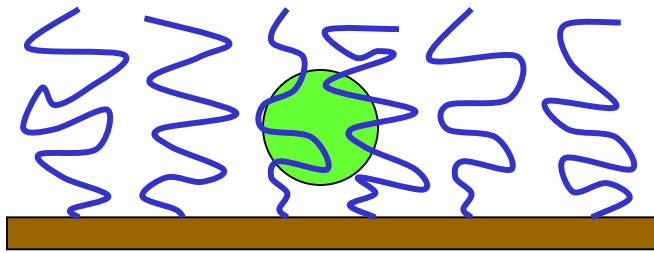
Insertion Into Brushes

A bird's-eye view

**A. Halperin
LIPHy - Grenoble**

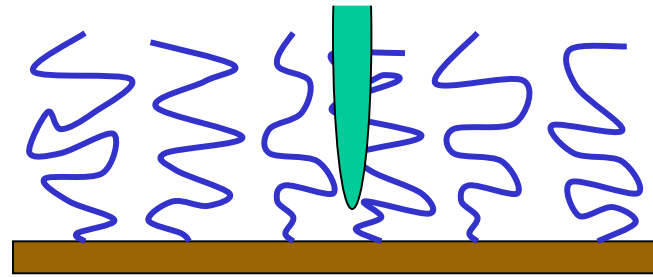
**G. Fragneto , A. Schollier - Grenoble
M. Sferrazza - Brussels
M. Kröger - Zurich
V. Ermilov, A. Lazutin - Moscow
E.B. Zhulina - St Petersburg**

INSERTION INTO BRUSHES: WHERE? WHAT?



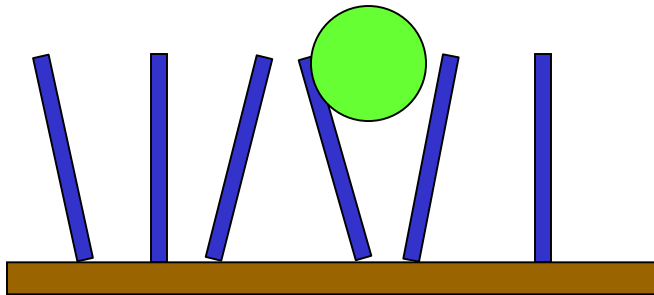
Colloids/Proteins/Cells

- Adsorption isotherm
- Concentration profiles
- T effects



AFM-sharp tips

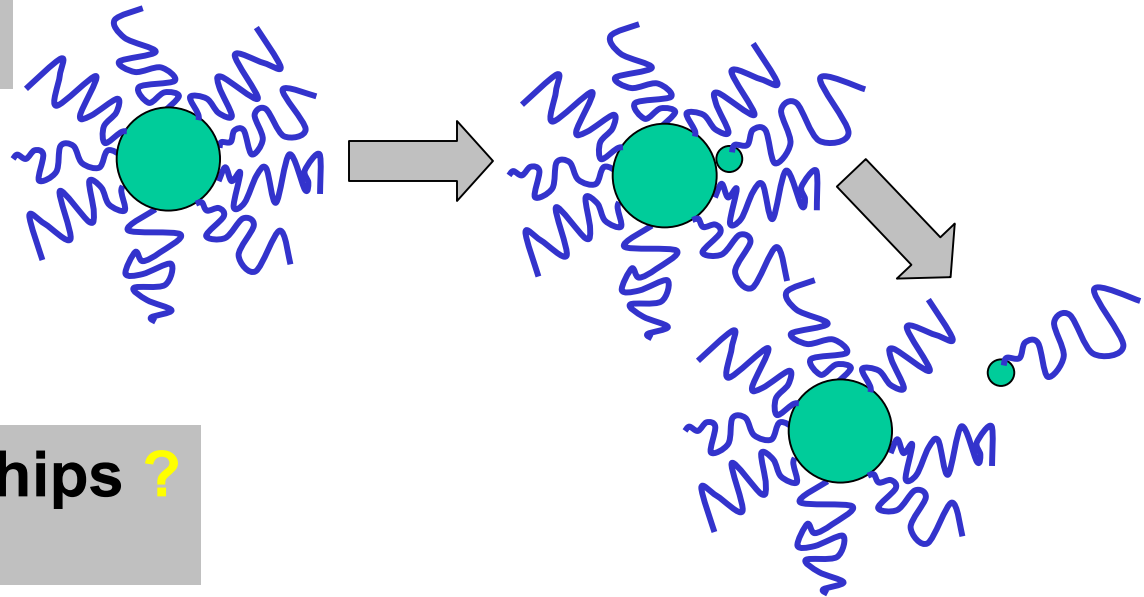
- Force laws



Protein Binding DNA Chips ?

- Adsorption Isotherms

Martha Bulik 1999...



Copolymer Exchange

- Rate Constants

So...

Need: **F** of particle

3 ingredients:

→ (1) Two limits: “Large” vs. “Small”

“Small” = Insertion \Rightarrow weak perturbation

→ (2) F_{ins} = work against unperturbed brush $\Pi(z)$
“universal penalty”

→ (3) Attraction & Adsorption modes
case specific

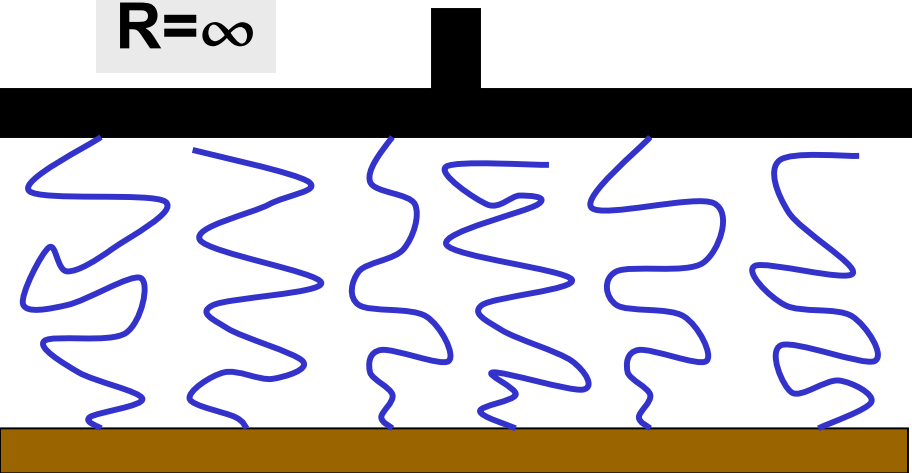
Then:

→ Some open questions

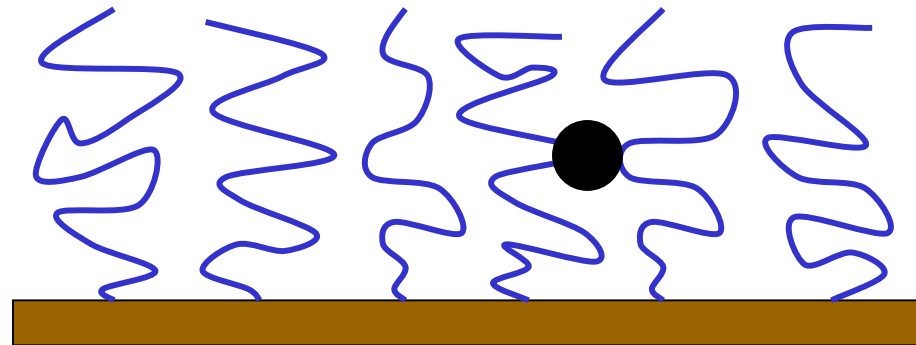
→ What did we actually do? \pm some details

Compressive vs. Insertive modes

$R = \infty$

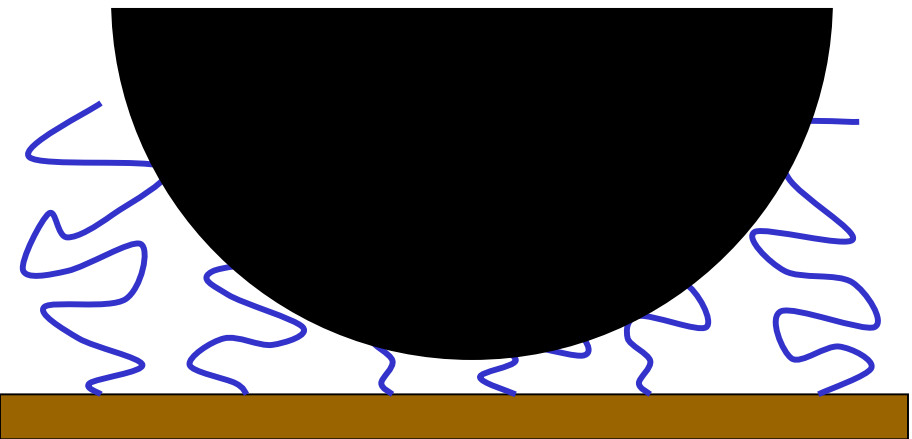


$R \approx a$



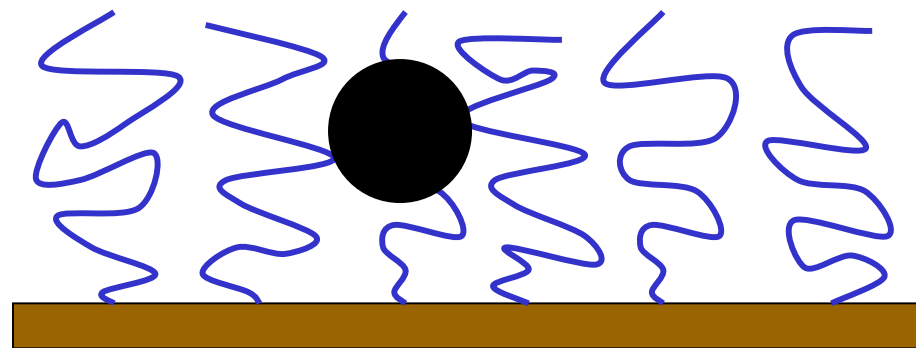
Compression mode

$H_0 \ll R < \infty$

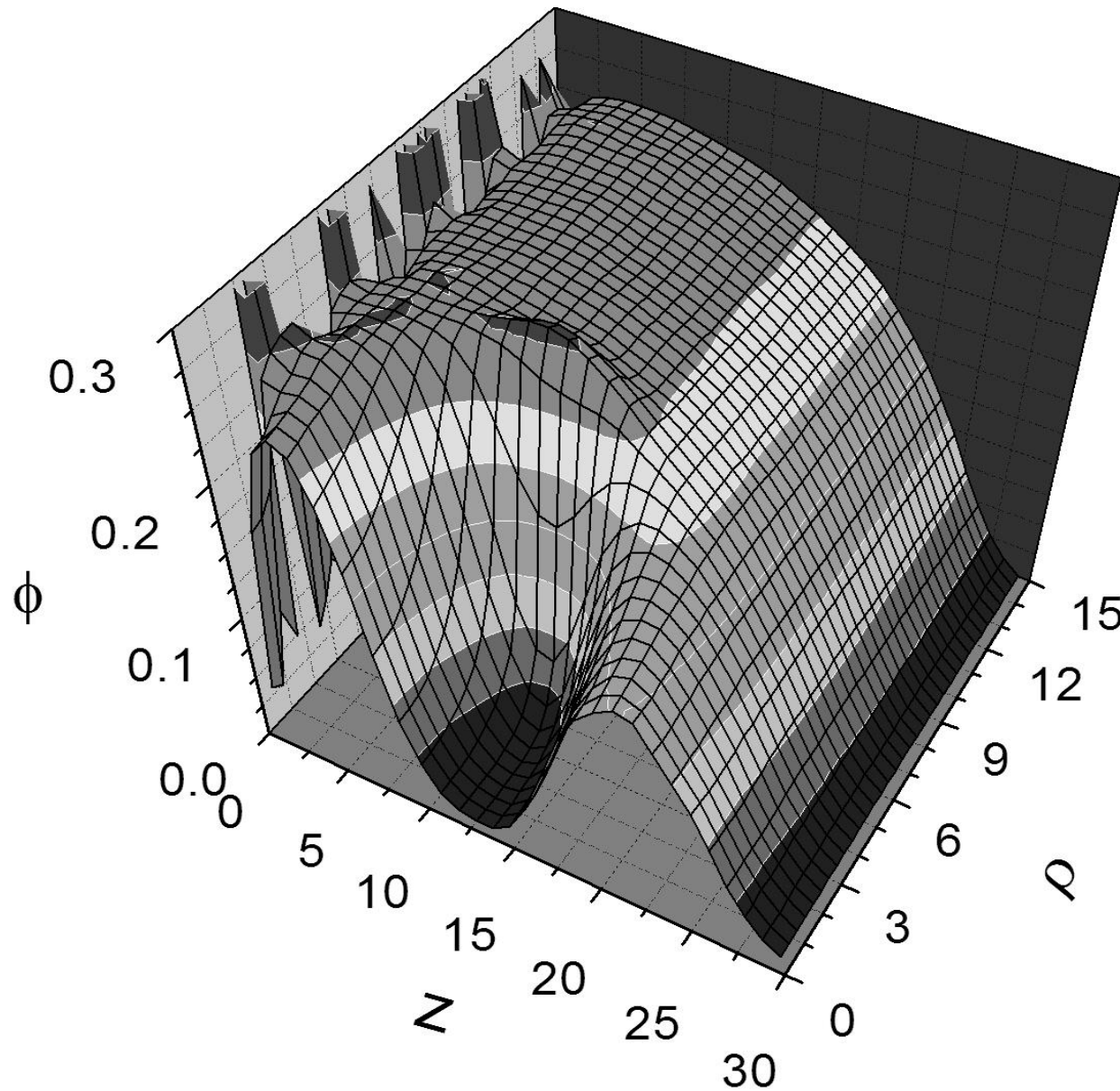


Insertion mode: short ranged
Perturbation of $\phi(z)$

$H_0 \gg R \gg a$



Insertion Mode-Concentration Profile I

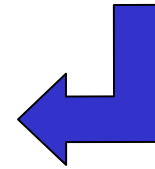


$R_p=4$
 $H=30$

Insertion limit

Small particle: circumvented by chain trajectories

$\Phi(z) \approx \Phi(z)$ of **unperturbed** layer



$\Phi(z)$ via Simulations, SCF etc

&

$\Pi(z) \approx \Pi(z)$ of **unperturbed** layer as determined by $\Phi(z)$



The insertion penalty

The Osmotic Insertion Penalty

Work against $\Pi(z) = \Pi(\phi(z))$ of the **unperturbed** brush

● AFM tip etc

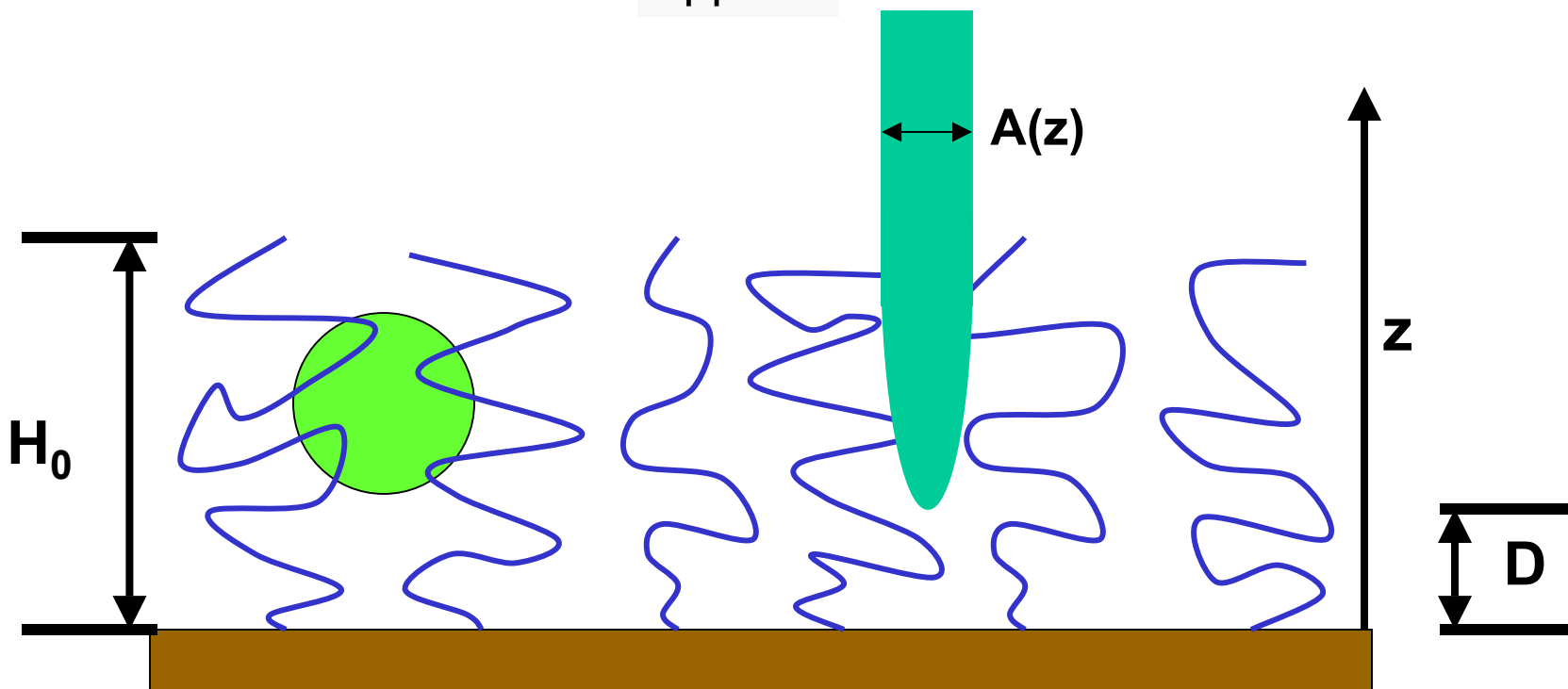
general

$$F_{ins} \approx \int_D^{H_0} \Pi(z) A(z) dz$$

● \approx sphere/globular protein

approx'

$$F_{ins} \approx \Pi(z) V_p$$

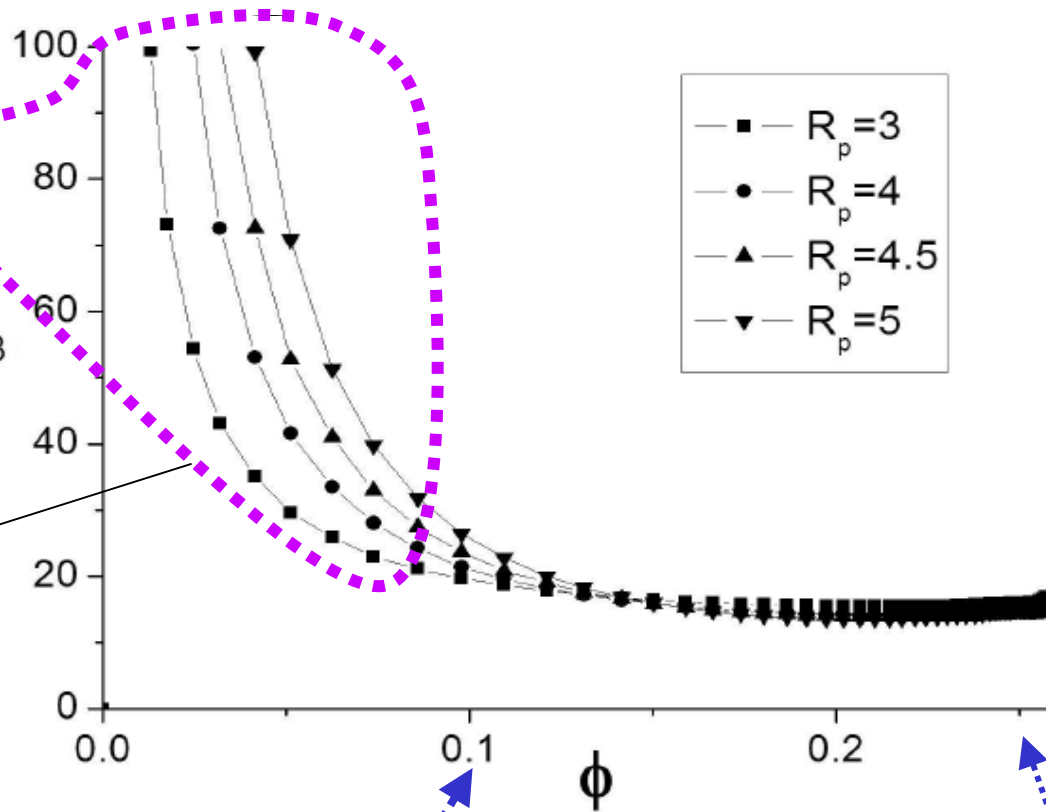


Insertion of a Sphere: Osmotic Penalty

$$\bar{R}_p = R_p + \frac{a}{2}$$

$$F_{ins} \left(\frac{\xi}{\bar{R}_p}\right)^3$$

Surface effects



$$z_{\max} \approx 35$$

$$z \approx 24$$

$$z \approx 4$$

$$F_{ins} \approx \Pi V_p \approx kT \left(\frac{R_p}{\xi}\right)^3$$

The Osmotic Penalty - Comments

Bulk vs. Brush

Brush

$\Pi(z)$ anisotropic

$$F_{\text{ins}}(z) \approx \Pi(z) V \quad R \ll H, z \ll H$$

Elastic & Interaction free energies

Bulk

Π isotropic

$$F_{\text{ins}} = \Pi V \quad \text{any } R, z$$

Interaction free energy

F_{ins} : effect of density

Decreasing Σ at constant T



Increase $\Pi(z)$ & F_{ins}

Brush collapse via T tuning at constant Σ



Decrease $\Pi(z)$ & F_{ins}

Σ area per chain

F_{ins}

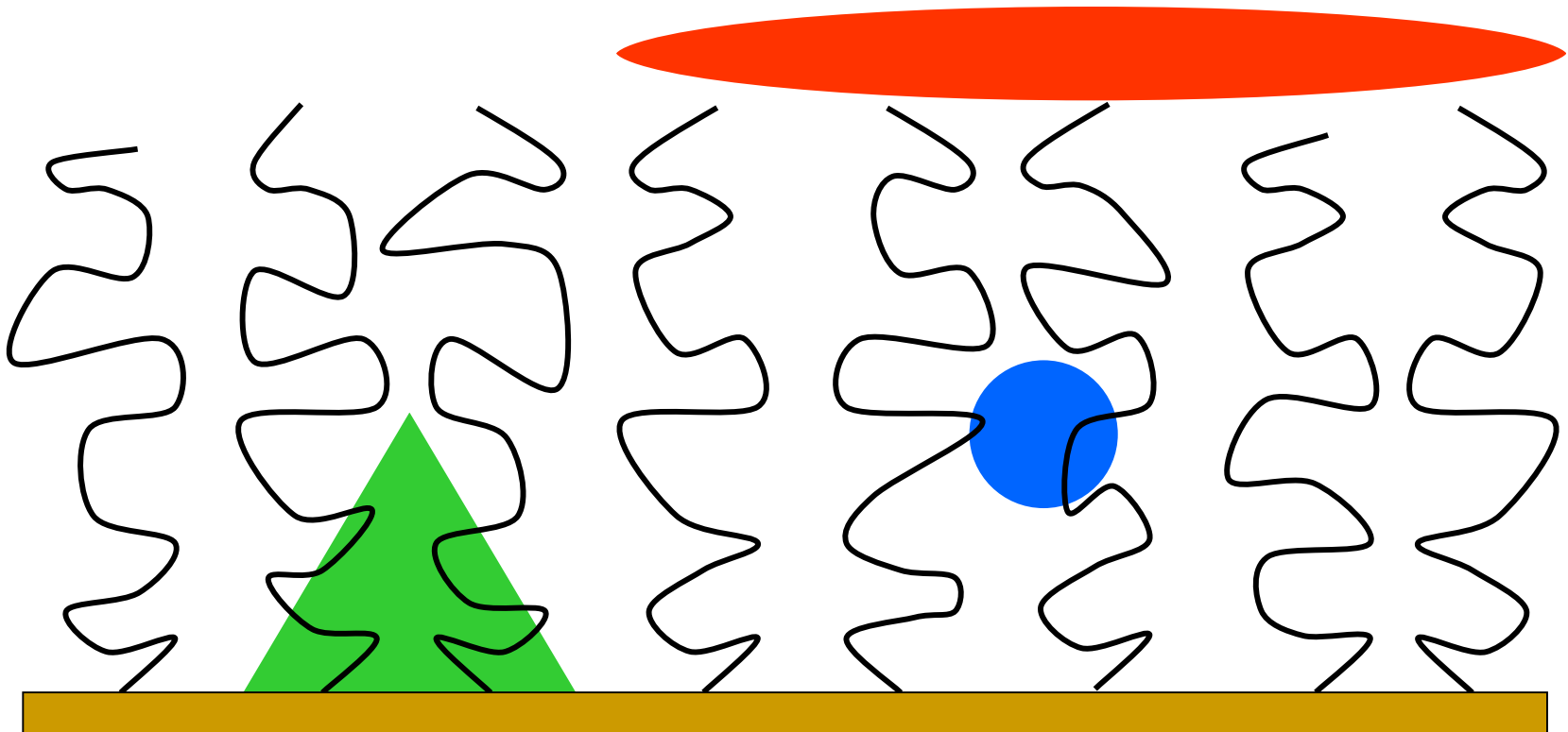


“Osmotic Archimedes Principle”

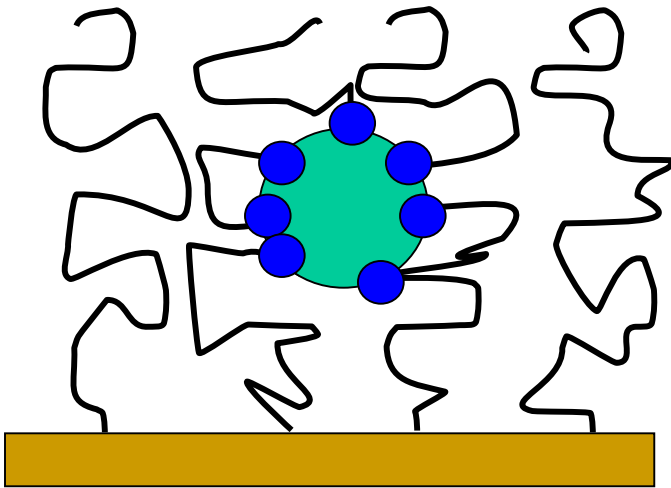
“Osmotic buoyancy”

Three Adsorption Modes

- **Primary Adsorption**-at the surface (Surface-Particle attraction)
- **Secondary Adsorption**-at the brush's outer edge (vdW attraction)
- **Ternary Adsorption**-within the brush (Polymer-Particle attraction)



Weak Non - Specific Ternary Adsorption



- no specific adsorption site
- uniform surface
- $-\epsilon kT$ per monomer in contact
- $\epsilon \ll 1$ below adsorption threshold
 - ▶ no bulk complexation
 - ▶ weak perturbation

$$F_{att} \approx -\epsilon kT c(z) A_p a$$

Proposed for proteins but more suitable for colloids

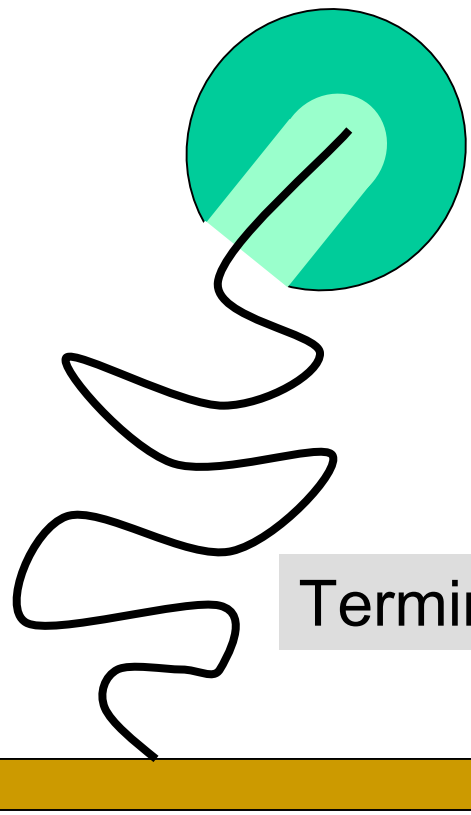
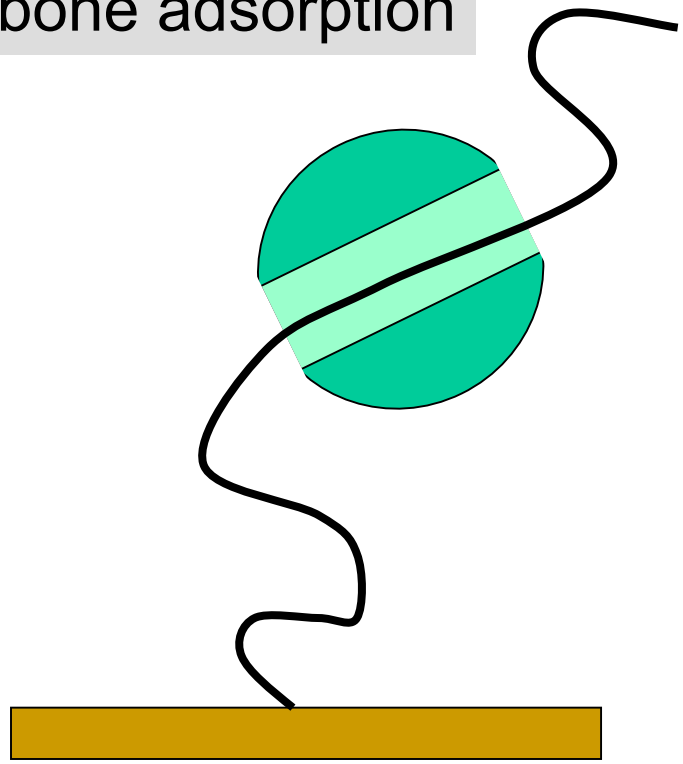
Strong Specific Ternary Adsorption: 2 Cases

$F_{\text{attraction}} \gg kT$
Specific Adsorption Site
Binds definite number of monomers

Strong Adsorption
 \Rightarrow Weak Perturbation

Antibodies - 2 types of binding sites:

Backbone adsorption



Terminal adsorption

The Insertion Mode

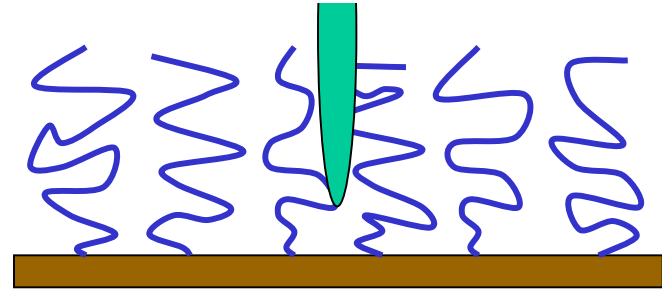
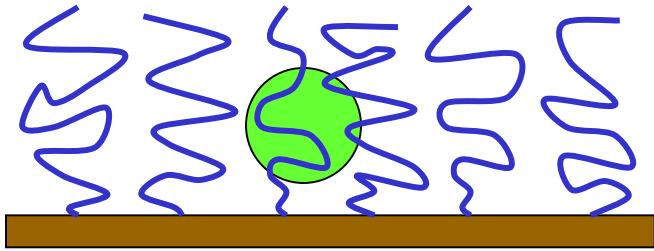
Open (?) Questions:

- What is small?
- The cross-over between small and big?
- Surface & edge effects?
- Shape effects?

In any case

**physically transparent & simple approximation
complementing
the “compression + Derjaguin’s” approximation
for “large” particles**

INSERTION INTO BRUSHES: 3 DIRECTIONS

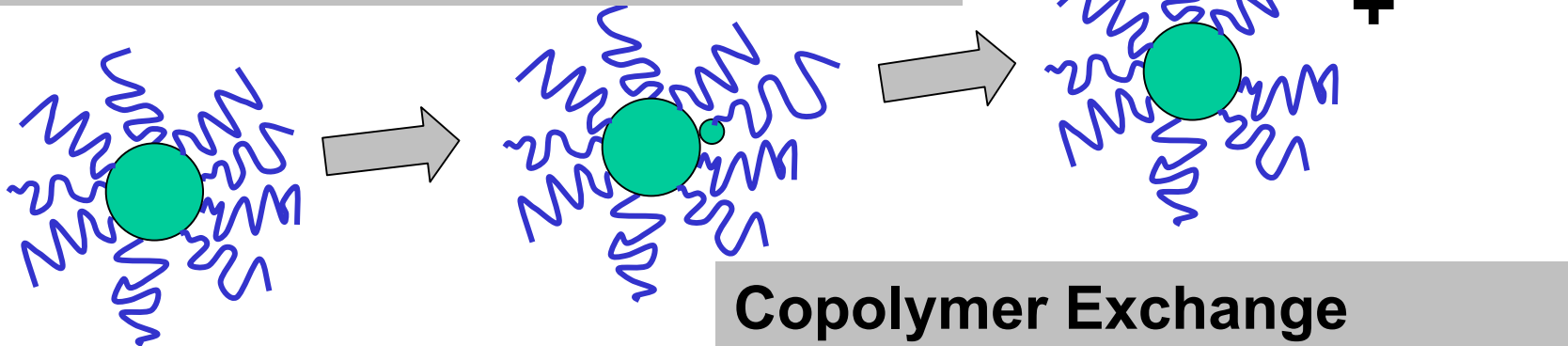


Colloids/Proteins (primary, ternary X3)

- Adsorption isotherms-SCF
- Concentration profiles-SCF
- F_{ins} - simulation: MC+Umbrella sampling
- Collapse effects-SCF
- (I) UCST - "classical"
- (II) LCST - "non classical" PNIPAM

AFM-sharp tips

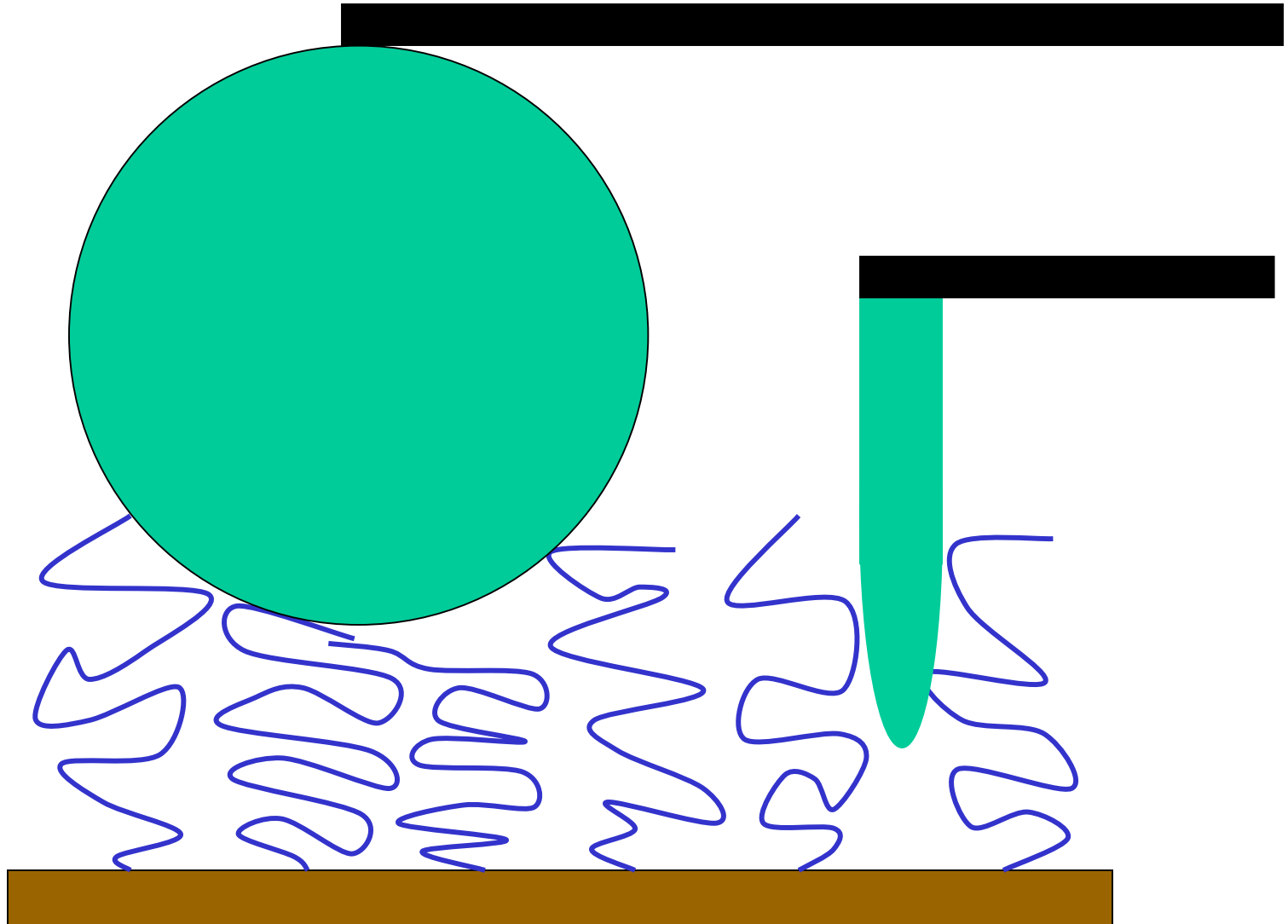
- Force laws



Copolymer Exchange

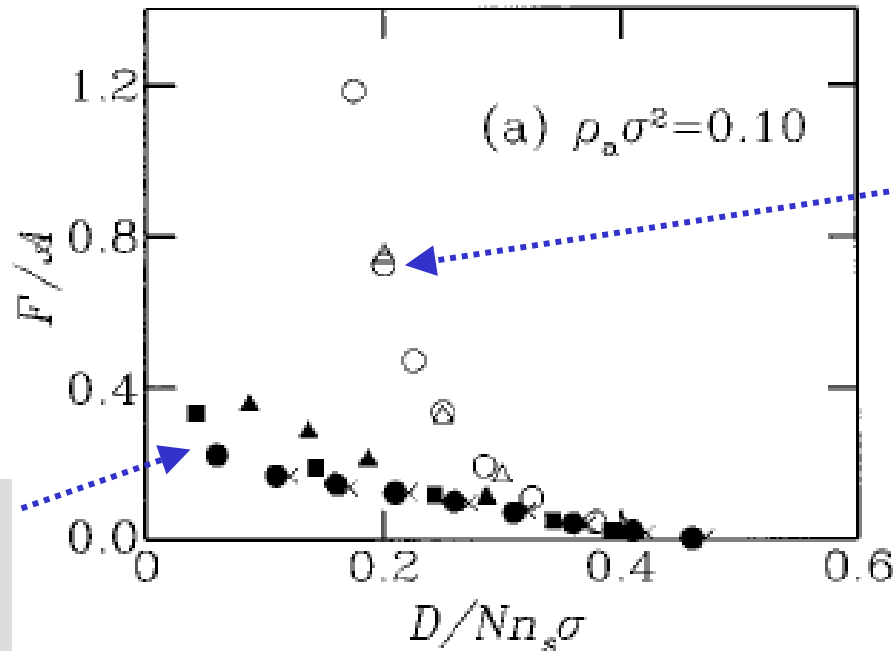
- Rate Constants
- Concentration Effects (slow down)

AFM force laws: Colloidal vs. Sharp tips



Simulation: Sharp Tip vs. “Planar Piston”

M. Murat & GS Grest, Macromolecules **1996**, 29, 8282



planar piston =
cylindrical tip within
Derjaguin's approx'

sharp tip:
 \approx cylindrical

Figure 1. Force-displacement curves for a brush with grafting density (a) $\rho_a = 0.1\sigma^{-2}$ and (b) $0.07\sigma^{-2}$ for chain length $N = 100$, compressed by an identical brush (\circ), an infinite, bare surface (Δ), and an AFM tip. The radii (r_{cyl}/σ , r_{sph}/σ) of the tip are (7, 10) (\times), (10, 14) (\bullet), (14, 20) (\blacksquare), and (16, 100) (\blacktriangle). The distance D between the surfaces is scaled by the chain length N and the number of surfaces bearing end-grafted polymer, n_s .

Sharp tips: (i) No compression (ii) weaker force law (iii) no theory

AFM: Insertion vs. Compression Force Law

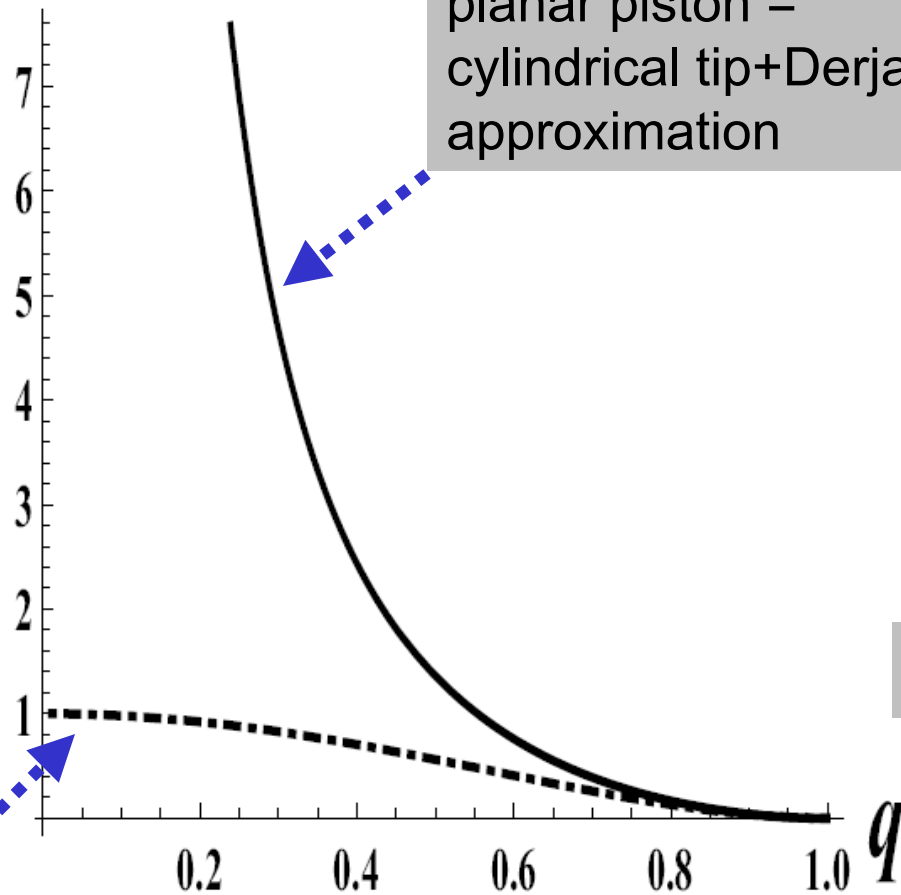
SCF concentration profile
marginal solvent

$$\frac{f}{A_{\text{cyl}} \Pi_0}$$

Murat & Grest parameters
 $\Pi_0 \approx 0.12kT/a^3$

cylindrical tip: osmotic
penalty

$$F_{\text{ins}} \approx \int_D^{H_0} \Pi(z) A(z) dz$$



planar piston =
cylindrical tip + Derjaguin's
approximation

$$q = D/H_0$$

AFM sharp tips vs. Colloidal tips & SFA

- SFA & Cylindrical AFM tip in the compression regime (compression + Derjaguin's approx')

SFA Experiments can be fitted by BOTH the Alexander model AND the SCF parabolic profile

- Cylindrical tip in the insertion regime: Osmotic Penalty

$$f_{ins} = A_{cyl} \Pi = \text{const}'$$

Alexander model

$$f_{ins} = A_{cyl} \Pi_0 \left(1 - \frac{D^2}{H_0^2} \right)^2$$

SCF parabolic profile

Qualitatively different predictions

→ AFM sharp tip force laws can distinguish between the 2 models

Repression of Primary Protein Adsorption

Two competing terms:

Why PEG works?

(1) Surface contact: U_{ads}

(2) Osmotic Penalty at the surface

$$\Pi \approx \frac{kT}{\xi^3}$$

&

$$\xi \approx \Sigma^{1/2}$$



$$\frac{F_{\text{ins}}}{kT} \approx \frac{V_{\text{protein}}}{\Sigma^{3/2}}$$

Adsorption isotherm:

$$\Gamma \approx K(\Sigma, N, V_{\text{protein}}) c_{\text{protein}}$$

$$K(\Sigma, N, V_{\text{protein}}, A_{\text{protein}}) = \exp\left(-\frac{\Delta F}{kT}\right)$$

$$\ln \Gamma \approx \ln c_{\text{protein}} + \frac{U_{\text{ads}}}{kT} - \frac{V_{\text{protein}}}{\Sigma^{3/2}}$$

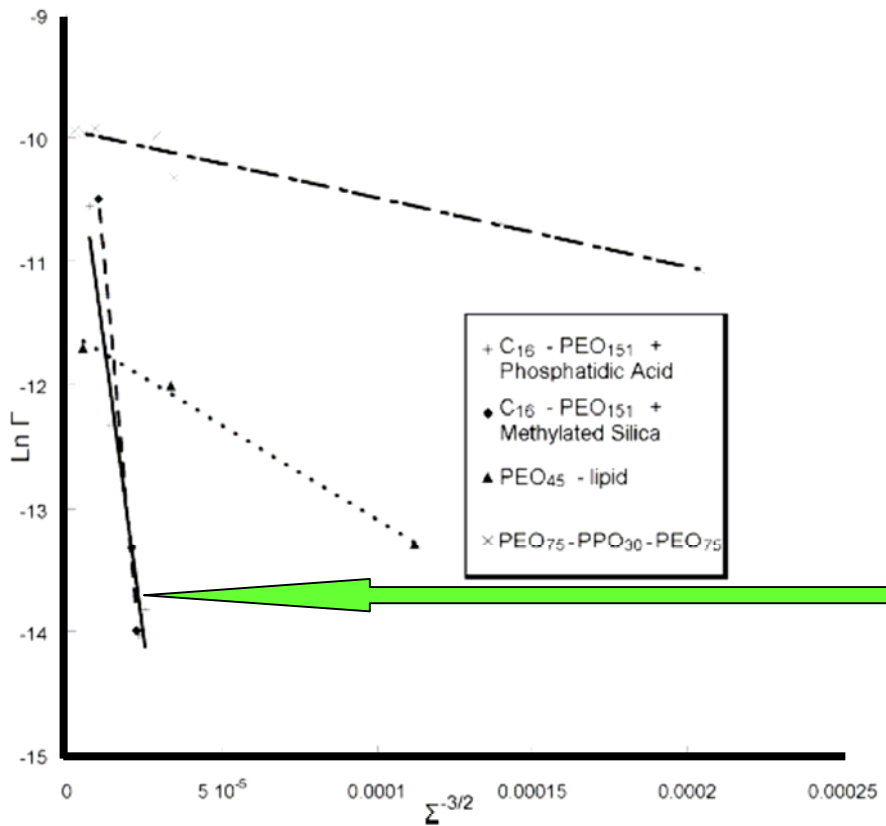


Dense brushes repress protein adsorption

Fibrinogen: Primary Adsorption?

Why PEG works?

$$\ln \Gamma \approx \text{const}' - \frac{V_{\text{protein}}}{\Sigma^{3/2}}$$



McPherson, T.; Kidane, A.; Szeifer, I.; Park, K. *Langmuir* 1998, 14, 176.

Malmsten, M.; van Alstine, J. M. *J. Colloid Interface Sci.* 1996, 177, 502–512.

Efremova, N. V.; Bondurant, B.; O'Brien, D. F.; Leckband, D. E. *Biochemistry* 2000, 39, 3441–3451.

fit:

$$V_{\text{protein}} \approx 2 \times 10^5 \text{ \AA}^3$$

literature: anhydrous fibrinogen

$$V_{\text{protein}} \approx 4.8 \times 10^5 \text{ \AA}^3$$